



Таким чином, в ході проведених випробувань було визначено ефективність флокулянту метацид при видаленні нафтопродуктів, що забезпечило ступінь вилучення 99%. Показано, що максимальну ефективність виявив флокулянт метацид при концентрації 2 мг/дм<sup>3</sup>.

### Література:

1. Domercq, P. Emission and fate modelling framework for engineered nanoparticles in urban aquatic systems at high spatial and temporal resolution / Prado Domercq, Antonia Praetorius, Alistair B. A. Boxall // *Environmental Science: Nano*. – 2018. – Vol. 5. – Iss. 2. – P. 533–543. doi: 10.1039/c7en00846e
2. Boxall, A. B. A. The environmental side effects of medication / A. B. A. Boxall // *EMBO reports*. – 2004. – Vol. 5. – Iss. 12. – P. 1110–1116. doi: 10.1038/sj.embor.7400307
3. Андрющенко, Е. Лекарства травят питьевую воду [Электронный ресурс] / Е. Андрющенко // *Днепр вечерний*. – 2017. – 17 июл. – Режим доступа: <http://dv-gazeta.info/vechyorka/zdorovje/lekarstva-travyat-pitevuyu-vodu.html>
4. Ермакович И. А., Самойленко Н. Н. Загрязнение муниципальных вод фармацевтическими препаратами и их производными // *Восточно-Европейский журнал передовых технологий*. — 2013. — № 64. — С. 8–11
5. Кульский Л.А., Строкач П.П. Технология очистки природных вод. – К.: Вища шк., 1981. – 328 с.
6. Шутько А.П., Басов В.П. Использование алюминийсодержащих отходов промышленных производств. – К.: Техника. – 1989. – 112 с.
7. Экстракционно-спектрофотометрический метод определения суммарного содержания нефтепродуктов в воде // *Химия и технология воды*. – 1999. – 21, №6. – С. 611–616.
8. Лурье Ю.Ю. Аналитическая химия промышленных и сточных вод. – М.: Химия, 1984. – 448 с.



UDC 544.723.2

## BIOSORBENTS FROM SUGARCANE RESIDUES

**V. Halysh<sup>1,2</sup>, O. Sevastyanova<sup>3,4</sup>, M. Gomelya<sup>1</sup>**

<sup>1</sup>*Igor Sikorsky Kyiv Polytechnic Institute,  
Peremogy Avenu 37, Kyiv, 03056, Ukraine*

<sup>2</sup>*O.O. Chuiko Institute of Surface Chemistry, National Academy of Sciences of Ukraine  
General Naumov St.17, Kyiv, 03164, Ukraine*

<sup>3</sup>*KTH Royal Institute of Technology  
Teknikringen 56-58, SE-100 44 Stockholm, Sweden*

<sup>4</sup>*KTH Royal Institute of Technology  
Teknikringen 56-58, SE-100 44 Stockholm, Sweden*

**e-mail:** v.galysh@gmail.com

In recent years, an increasing interest in obtaining of sorbents based on renewable lignocellulosic plant sources is observed. Plant raw materials can yield efficient sorbents of inorganic and organic pollutants for water treatment. Agricultural residues and waste from the food industry are a promising source of biosorbents due to the presence of pores and of



functional groups on the surface [1, 2]. The main advantages of biosorbents over other traditional adsorbents are their annual renewal and low cost. Different researchers have investigated the sorption properties of various plant materials for heavy metals and for organic dyes [3-6].

Biosorbents can be chemically modified in order to obtain products with a larger number of functional groups and active sites. Various chemicals have been proposed, for example, organic and mineral acids, alkaline solutions, combined treatments, organic compounds, etc [7-10]. Solutions of acid are strong oxidizing agents that lead to polysaccharide hydrolysis and to the corrosion of equipment. Treatment with alkaline solution promotes the formation of additional active sorption sites but does not promote the increase in the surface area. The disadvantages of combined treatment are the multi-stage and the large water consumption. The application of organic compounds is connected with the use of additional chemicals. A promising method for obtaining biosorbents is the oxidative treatment of plant materials with hydrogen peroxide in acetic acid [11]. It is possible to obtain biosorbents with different contents of cellulose and lignin by varying the concentration of hydrogen peroxide. This leads to the oxidation and partial dissolution of lignin, and promotes the oxidation of polysaccharides into soluble products, and a porous structure is formed.

The aim of the research work was to study the effect of oxidative treatment on the structural and sorption properties of biosorbents from sugarcane residues.

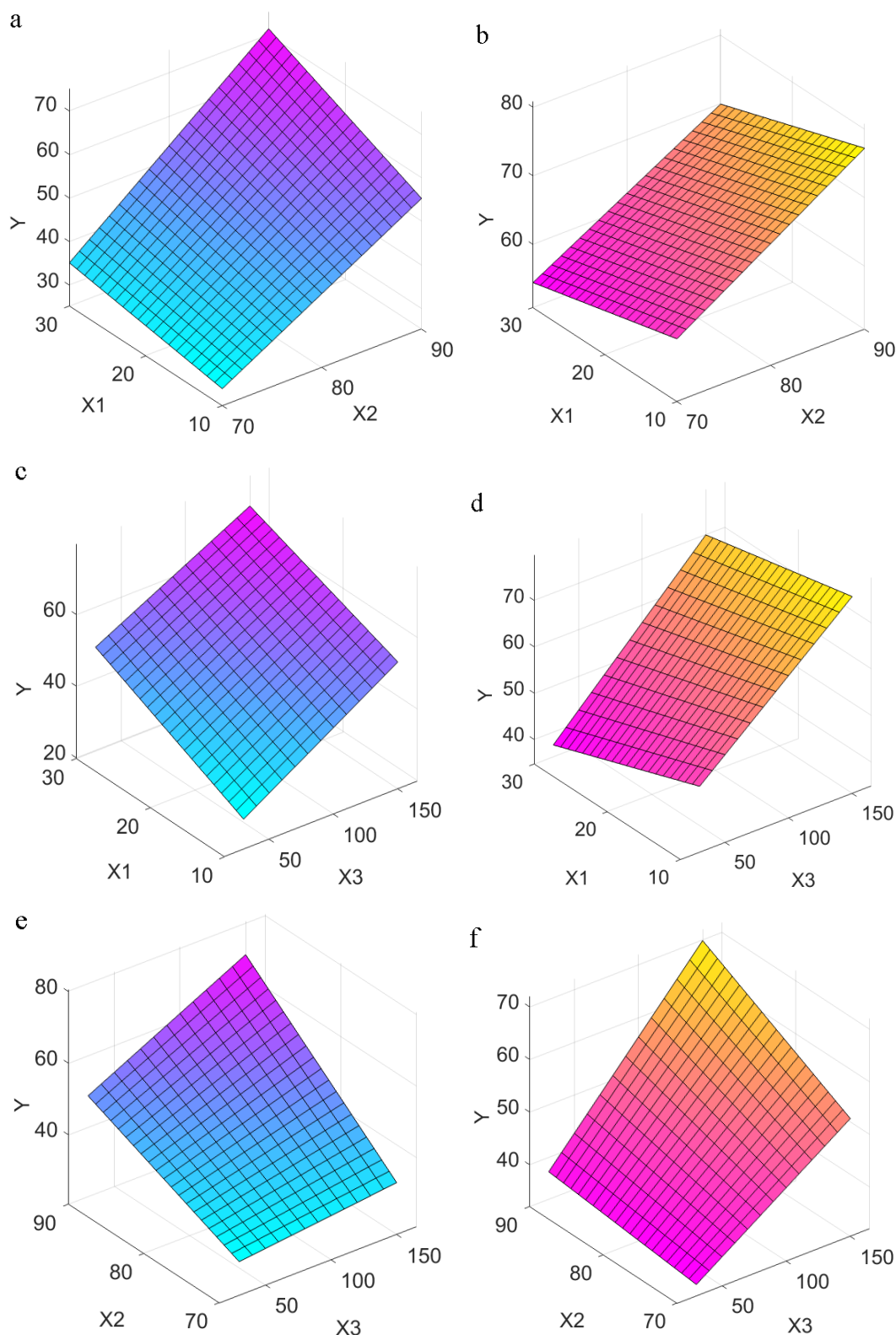
The residues of sugarcane processing, i.e. bagasse and straw (tips and leaves), were used as a raw material. To prepare biosorbents, plant materials were treated with hydrogen peroxide in an acetic acid medium at different hydrogen peroxide content in solution ( $X_1$ ), temperature ( $X_2$ ) and time of the process ( $X_3$ ). The yield, chemical composition of obtained materials and their sorption ability towards methylene blue was studied. The results of the investigation of efficiency of cationic dye removal from water solution are given as 3D-surfaces in Fig. 1.

Both sugarcane biomasses were easily delignified by hydrogen peroxide treatment in an acetic acid medium. Such treatment causes the removal of 47.9-53.7% material and results in the preparation of biosorbents of a white color indicating a low lignin content due to the delignification of biomass with peracetic acid which is formed as a result of an interaction of acetic acid with hydrogen peroxide. At the same time, the removal of different extractives also takes place.

Partial removal of the components of polysaccharide nature can also takes place. The yield and chemical composition of the biosorbents strongly depends on the treatment parameters as well as their sorption ability.

The greatest removal efficiency of methylene blue from water was 77.2% for a straw sorbent obtained by treating biomass with hydrogen peroxide of 10 vol. % at a high temperature for a long time. This can be due to the removal of most part of mineral components from straw that promotes the formation of a porous structure. It can be seen that increasing the hydrogen peroxide content, temperature and time leads to increase in the sorption properties of biosorbents to cationic dye.

The greatest efficiency of methylene blue sorption by both types of biosorbents is achieved after 150 min treatment at 90°C with a hydrogen peroxide concentration of 10 vol.%. But, in case of biosorbent from straw, the obtained material shows the greatest sorption of organic dye after sugarcane straw treatment with a hydrogen peroxide concentration of 30 vol.%.



**Figure 1. Surface plots of the effects of interactions on sorption ability of biosorbents from sugarcane biomass towards methylene blue:**

- a, b** –  $X_1$ - $X_2$  /  $H_2O_2$  content (vol. %) - temperature ( $^{\circ}C$ ) for bagasse and straw, respectively;  
**c, d** –  $X_1$ - $X_3$  /  $H_2O_2$  content (vol. %) – time (min) for bagasse and straw, respectively;  
**e, f** –  $X_2$ - $X_3$  / temperature ( $^{\circ}C$ ) – time (min) for bagasse and straw, respectively



Thus, the efficient biosorbents was prepared from sugarcane residues. The treatment of s bagasse and straw with hydrogen peroxide gives cellulosic sorbents with different structural and sorption properties depending on the ratio of reagents. Comparison of the initial materials and prepared biosorbents demonstrated that the sorption ability of the both materials toward cationic dye depends on the treatment parameters.

#### References:

1. Anastopoulos I., Kyzas G.Z., Agricultural peels for dye adsorption: A review of recent literature. *Journal of Molecular Liquids*. – 2014. – №200. – P.381–389.
2. Halysh V., Sevastyanova O., Riazanova A.V., Pasalskiy B., Budnyak T., Lindström M.E., Kartel M. Walnut shells as a potential low-cost lignocellulosic sorbent for dyes and metal ions. *Cellulose*. – 2018. – №25(8). – P. 4729–4742.
3. Suteu D., Zaharia C., Badeanu M. Agriculture wastes used as sorbents for dyes removal from aqueous environments. *Lucrari Stiintifice*. – 2010. – №53(1). – P. 140–145.
4. Kartel M., Galysh V. New composite sorbents for caesium and strontium ions sorption. *Chemistry Journal of Moldova*. – 2017. – №12(1). – P. 37–44.
5. Li K., Wang Q., Dang Y., Wei H., Luo Q., Zhao F. Characteristic and Mechanism of Cr(VI) Biosorption by Buckwheat Hull from Aqueous Solutions. *Acta Chimica Sinica*. – 2012. – №70(7), – P. 929–935.
6. Suteu D., Zaharia C., Badeanu M. Agriculture wastes used as sorbents for dyes removal from aqueous environments. *Lucrari Stiintifice*. – 2010. – №53(1). – P. 140–145.
7. Németh D., Barta Z., Labidi J., Gubicza L., Bélafi-Bakó K. Study on sorption characteristics of citric acid modified rape-seed pellet considering the chemical pre-treatment processes. *Hungarian Journal of Industrial Chemistry*. – 2011. – №39(3). – P. 407–411.
8. Shah J., Jan M.R., Haq A.U., Sadia M. Biosorption of cadmium from aqueous solution using mulberry wood sawdust: equilibrium and kinetic studies. *Sep Sci Technol*. – 2011. – №46(10), – P. 1631–1637.
9. Yu L.-X., Wang L.-Y., Chi R.-A., Zhang Y.-F., Xu Z.-G., Guo J. Competitive adsorption of  $Pb^{2+}$  and  $Cd^{2+}$  on magnetic modified sugarcane bagasse prepared by two simple steps. *Appl Surf Sci*. – 2013. – №268. – P. 163–170.
10. Kumar P.S., Ramalingam S., Abhinaya R.V., Kirupha S.D., Murugesan A., Sivanesan S. Adsorption of metal ions onto the chemically modified agricultural waste. *Clean – Soil, Air, Water*. – 2012. – №40. – P. 188–197.
11. Halysh V., Trembus I., Deykun I., Ostapenko A., Nikolaichuk A., Initska G. Development of effective technique for the disposal of the *Prunus Armeniaca* seed shells. *Eastern-European Journal of Enterprise Technologies*. – 2018. – №1(10). – P. 4–9.

